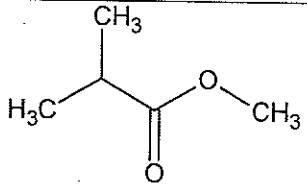
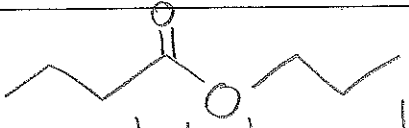


Answer key

Predict the products

1.	2-hexene + F ₂ →	2,3-difluorohexane
2.	3-heptene + H ₂ O $\xrightarrow{\text{acid}}$	2-heptanol, 3-heptanol
3.	3-hexanol $\xrightarrow{\text{H}_2\text{SO}_4}$	3-hexene
4.	propane + Cl-Cl $\xrightarrow{h\nu}$	chloropropane
5.	 + H-OH $\xrightarrow{\text{NaOH}}$	
6.	HCl + 3-pentene →	2-chloropentane, 3-chloropentane
7.	2-hexanol + Cr ₂ O ₇ $\xrightarrow{\text{acid}}$	2-hexanone
8.	ethanoic acid + N-ethyl, methanamine →	N-ethyl-N-methylethanamide + water
9.	1-pentanol + Cr ₂ O ₇ $\xrightarrow[\text{heat under reflux}]{\text{acid}}$	pentanoic acid
10.	H ₂ + 1-butene →	butane
11.	1-octanol + Cr ₂ O ₇ $\xrightarrow[\text{distill}]{\text{acid}}$	octanal

12. heptane + O ₂ →	carbon dioxide + water
13. Incomplete combustion of propane	carbon monoxide
14. propanol + butanoic acid	$\xrightarrow[\Delta]{\text{H}_2\text{SO}_4}$  propyl butanoate + water
15. H ₂ C=CH ₂ + O ₂ →	carbon dioxide + water

16. Which of the above are addition reactions?

1, 2, 6, 10

17. Which of the above are condensation reactions?

8, 14

18. Give specific industrial uses of the following reactions:

- Hydration - production of alcohols
- Addition of H₂ - make veg. oil solid (margarine)
- Condensation of alcohol and carboxylic acid - form esters - artificial flavourings

19. List 4 uses of esters

- perfume
- artificial flavours
- polyesters → plastics
- solvents

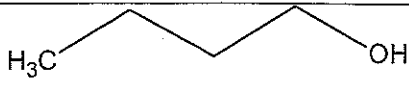
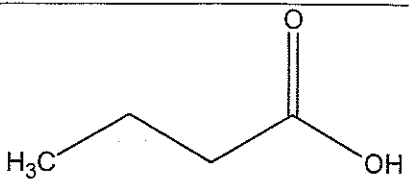
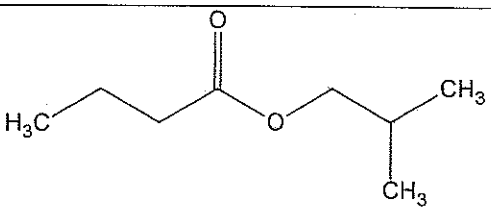
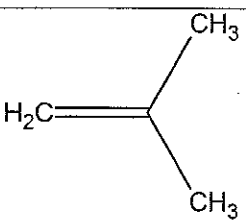
20. Describe a test that can be done to identify the presence of

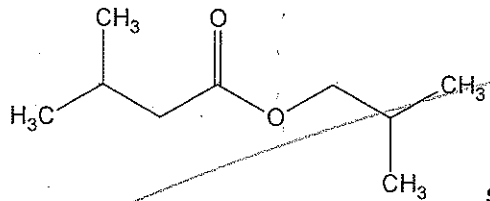
a. An alkene

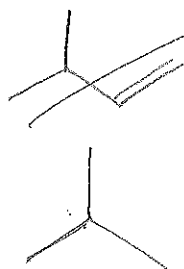
- add Br₂ if colour disappears, it is an alkene

b. An Alcohol

Retrosynthetic analysis: Give the reactants that could be used to make the following

1.		1-bromobutane + NaOH
2.		1-butanol + $K_2Cr_2O_7 \xrightarrow[\Delta]{\text{acid}}$
3.		butanoic acid + 2-methyl-1-propanol in acidic solution + heat
4.		1-bromo-2-methylpropane + NaOH Δ

5. Describe (giving specific reactions and the conditions under which the reactions would occur) how you could make  starting with alkenes.



requires preparation of non-markovnikov product \rightarrow not required

